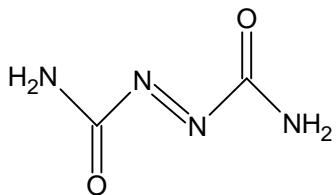


産業技術総合研究所 高エネルギー物質研究グループ
発熱分解エネルギー測定の標準化 热分析結果

Azodicarbonamide

$\text{H}_2\text{NCON=NCONH}_2$

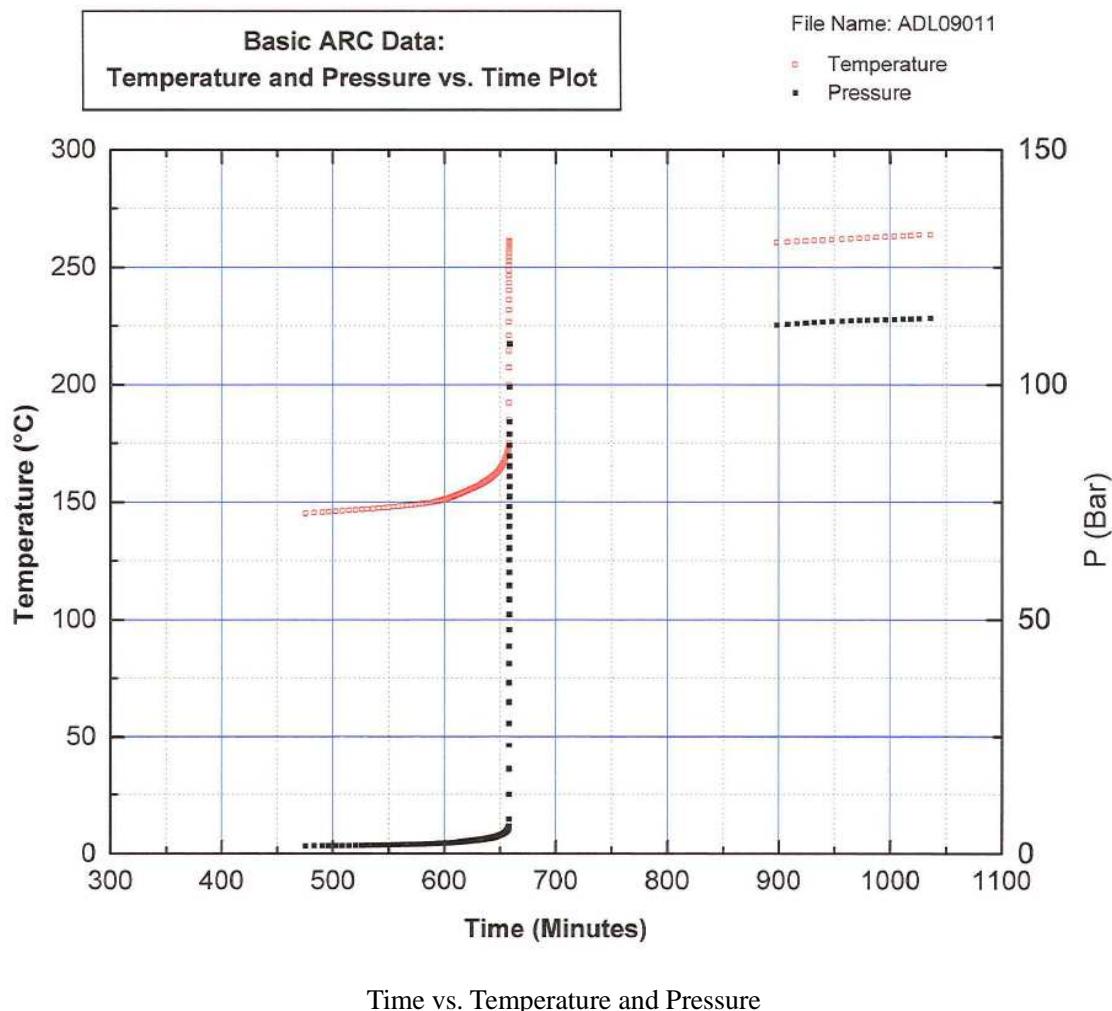
ADCA

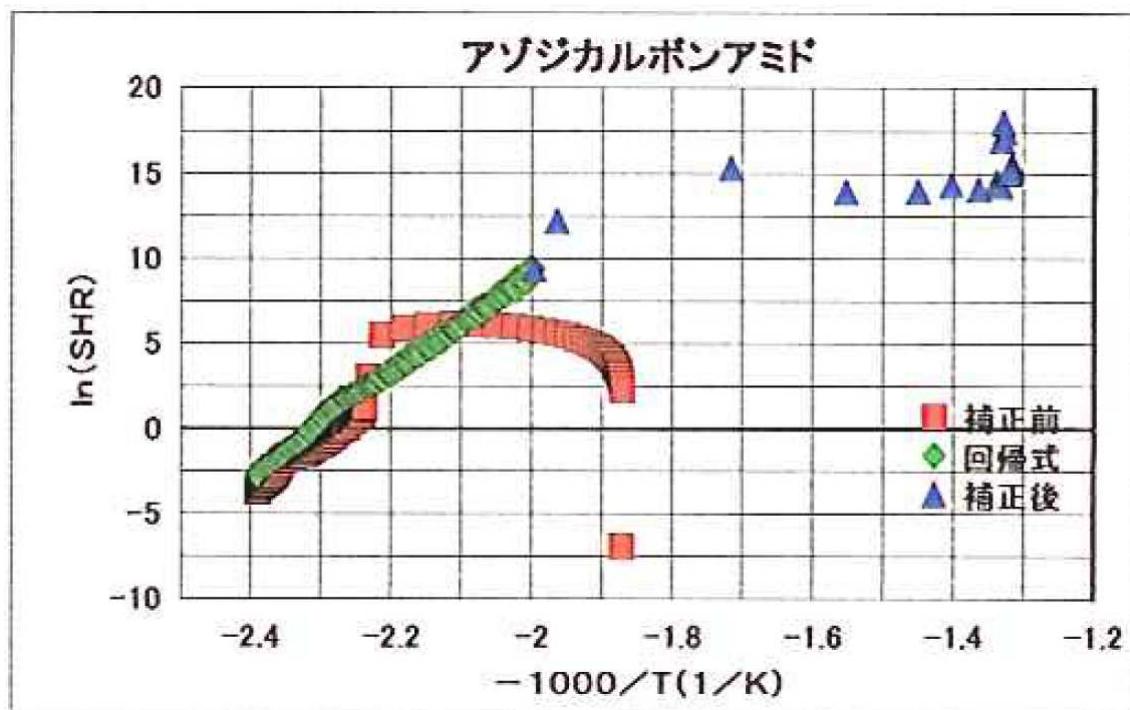


ARC device: ARC2000 (Arthur D. Little Inc.)

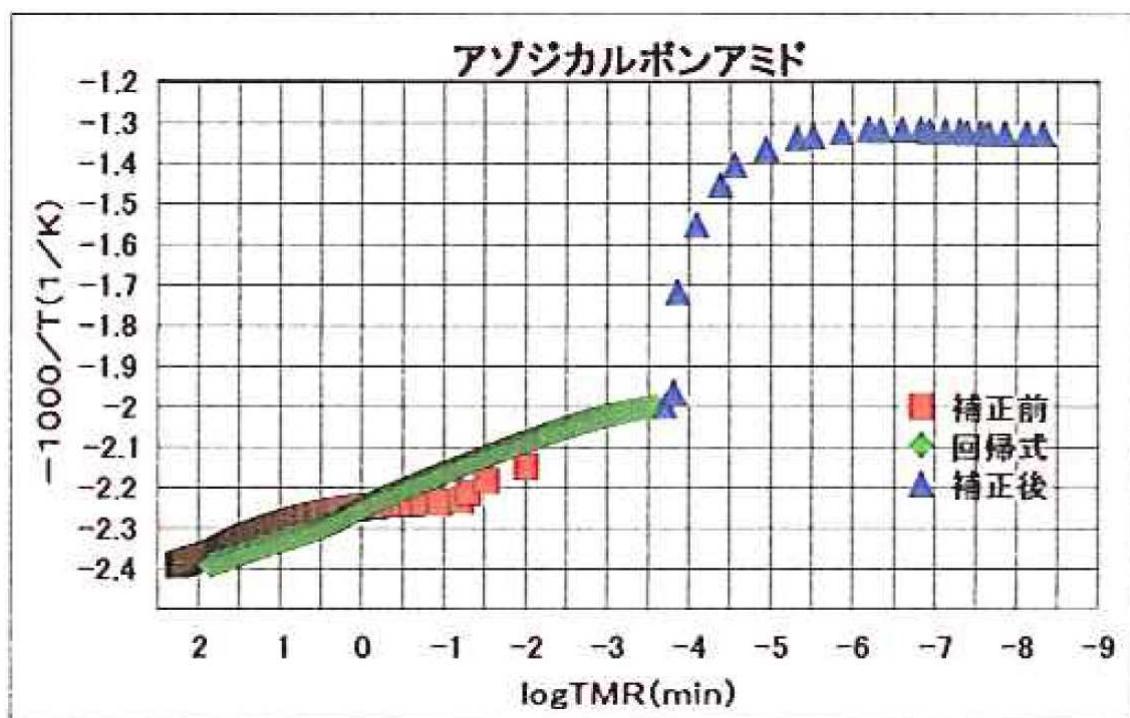
Date: 2009/1

Operator: KJ

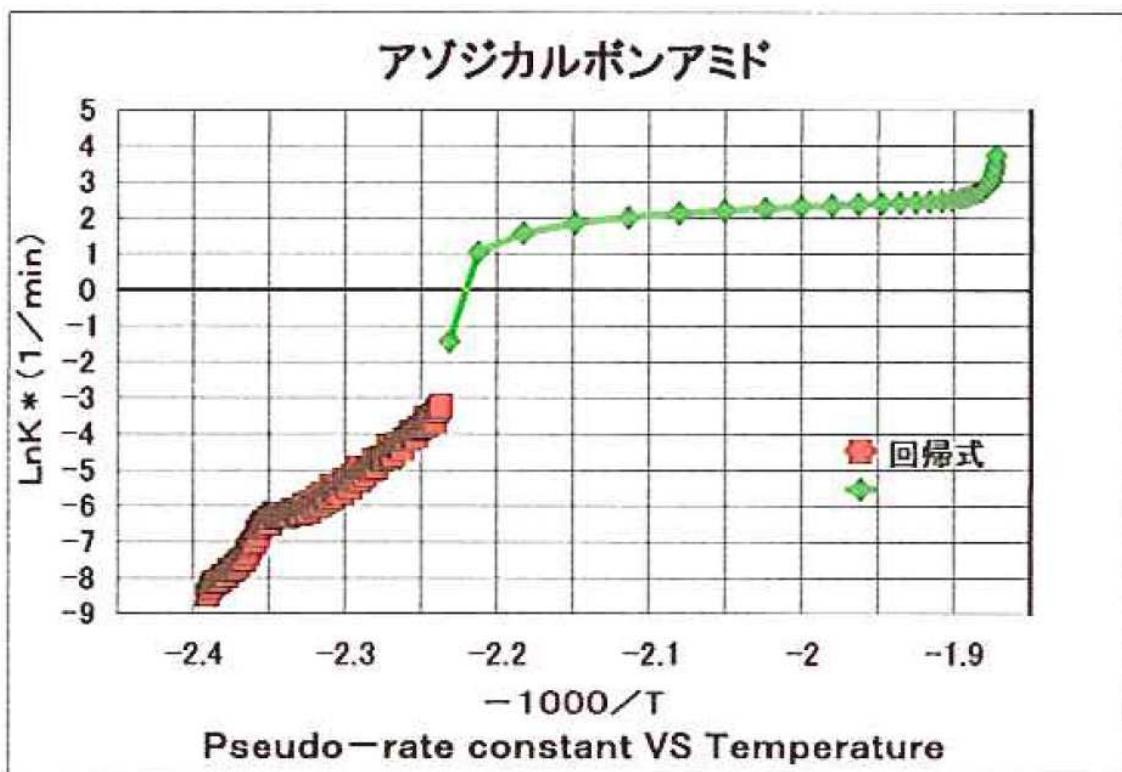




Temperature vs. Self heating rate



TMR vs. Temperature



Arrhenius equation (approximate calculation)

	Date	2009/1/15
Measuring conditions	ARC device	ARC2000 (Arthur D. Little Inc.)
	Operating Institute	KJ
	Operator	KJ
	Material of Bomb	Hastelloy C
	Weight of Bomb (g)	15.120
	Volume of Bomb (mL)	about 9
	Weight of sample (g)	1.654
	Weight of residue (g)	0.510
	Specific heat of Bomb ($J K^{-1} g^{-1}$)	0.419
	Specific heat of sample ($J K^{-1} g^{-1}$)	2.093
	ϕ facotr	2.83
	Start temperature (°C)	50
	End temperature (°C)	350
	Temperature increment (K)	5
	Waiting time (min)	10

	Searching time (min)	10
	Exothermic threshold (K min^{-1})	0.02
	Logging intervals ($^{\circ}\text{C}$)	0.2
	Pressure limit (kPa)	17000
	Atmosphere	Air, atmospheric pressure
Results	T_o , Exothermic temperature ($^{\circ}\text{C}$)	145.13
	Self heating rate at T_o (K min^{-1})	0.024
	Pressure at T_o (kPa)	170
	Temperature at maximum self heating rate ($^{\circ}\text{C}$)	199.96
	Maximum self heating rate (K min^{-1})	465.15
	Pressure at maximum self heating rate (kPa)	2790
	Pressure rising rate at maximum self heating rate (kPa min^{-1})	29107
	Maximum pressure (kPa)	10870
	Maximum pressure rising rate (kPa min^{-1})	33107
	Temperature at maximum pressure rising rate ($^{\circ}\text{C}$)	291.07
	Time to maximum rate (min)	182.64
	Maximum temperature ($^{\circ}\text{C}$)	261.27
	Adiabatic temperature rise ($^{\circ}\text{C}$)	116.14
	Activation energy (kJ mol^{-1})	250.8
	Heat of decomposition (J g^{-1})	686.6
Corrected results	T_{ARC} , Exothermic temperature ($^{\circ}\text{C}$)	135.66
	Time of maximum rate at T_{ARC} (min)	318.85
	Self heating rate at T_{ARC} (K min^{-1})	0.02
	Maximum self heating rate (K min^{-1})	6.44×10^7
	Maximum temperature ($^{\circ}\text{C}$)	486.96
	Adiabatic temperature rise ($^{\circ}\text{C}$)	351.30
	Heat of decomposition (J g^{-1})	736.9